

# Answer Key

## Topic 3: States of Matter

### Set 1: Multiple Choice Questions

1. What is the scientific word for what everything is made of?

**Answer: C) Matter**

2. Which state of matter has particles that are packed very close together and arranged in order?

**Answer: A) Solid**

3. What process occurs when a liquid is heated and turns into a gas?

**Answer: D) Evaporation**

4. What is a substance in its simplest form, such as gold, called?

**Answer: C) Element**

5. Which of the following describes a reversible, physical change?

**Answer: B) Melting ice**

6. Why can powders sometimes be confused for liquids?

**Answer: C) They can be poured into a container**

7. What happens to particles in a solid?

**Answer: C) They are always vibrating**

8. What is formed when two or more elements are combined, such as hydrogen and oxygen making water?

**Answer: A) A compound**

9. What is the reverse process of melting?

**Answer: C) Solidifying or freezing**

10. What is created when plaster of Paris powder is mixed with water to make a hard cast for a broken bone?

**Answer: B) A chemical reaction**

## Set 2: True or False

11. The particles in a solid do not move or vibrate at all.

**Answer: False (They are always moving and vibrating, but cannot move around from place to place)**

12. A chemical reaction occurs when particles are rearranged to make a new product.

**Answer: True**

13. The particles in a gas have very small spaces between them.

**Answer: False (They have large spaces between them)**

14. Condensation happens when the temperature is lowered and a gas turns back into a liquid.

**Answer: True**

15. Digestion of food in our bodies is an example of a chemical reaction.

**Answer: True**

16. Powders are crushed solids that have changed their state into liquids.

**Answer: False (They are crushed solids and have not changed their state)**

17. Changes of state between solids, liquids, and gases are irreversible.

**Answer: False (They are physical changes and are reversible)**

18. A single grain of sugar keeps its shape because it is a solid.

**Answer: True**

19. Some solids need more heat than others to make them melt.

**Answer: True**

20. Plastic is a useful product made from oil through a chemical reaction that cannot be reversed.

**Answer: True**

## Set 3: Fill in the Blanks

21. The smallest parts of a substance that are unique to that substance are called \_\_\_\_\_.

**Answer: particles**

22. Scientists use a \_\_\_\_\_ model to help explain the differences between solids, liquids, and gases.

**Answer: particle**

23. When temperature is increased, a solid will change into a liquid in a process called \_\_\_\_\_.

**Answer: melting**

24. Chemical changes are different from physical changes because they cannot be reversed; we say the changes are \_\_\_\_\_.

**Answer: irreversible**

25. A change of state from a liquid to a solid is called solidifying or \_\_\_\_\_.

**Answer: freezing**

26. Water is a combination of hydrogen and oxygen, which means it is an example of a \_\_\_\_\_.

**Answer: compound**

27. The particles in a \_\_\_\_\_ fill the shape of the container they are in and can move around each other.

**Answer: liquid**

28. To reverse evaporation, the temperature needs to be lowered to cause \_\_\_\_\_.

**Answer: condensation**

29. Five different types of \_\_\_\_\_ enquiry include research, fair testing, observing over time, identifying and classifying, and pattern seeking.

**Answer: scientific**

30. In a solid, the particles are packed closely together and are arranged in a \_\_\_\_\_ way.

**Answer: regular**

## Set 4: Match the Term

31. The process that is the exact reverse of freezing.

**Answer: Melting**

32. A type of matter where particles are arranged in a random way, cannot be squashed, and touch at least one other particle.

**Answer: Liquid**

33. Substances that are made of only one type of element in its simplest form, like gold.

**Answer: Elements**

34. A material, also known as gypsum, which forms a hard cast through a chemical reaction when mixed with water.

**Answer: Plaster of Paris**

35. The state of matter where particles are the most spread out with large spaces between them.

**Answer: Gas**

36. An action where you squeeze something to make it smaller or take up less space (which liquids and solids generally cannot do).

**Answer: Compress**

37. A type of reaction that creates a new product and cannot be reversed back to its original state.

**Answer: Chemical reaction / Chemical change**

38. The state of crushed solids, like sugar or sand, that can be poured into a container but still remain solids.

**Answer: Powder**

39. The physical process where a gas is cooled and returns to its liquid state.

**Answer: Condensation**

40. The overarching term for what metals, plastics, ceramics, glass, and fabrics are classified as.

**Answer: Materials**

## Set 5: Short Answer Questions

41. According to the text, what is the difference between an element and a compound?

**Answer: An element is a substance in its simplest form, while a compound is made from two or more elements joined together.**

42. How are the particles in a liquid different from the particles in a solid?

**Answer: The particles in a liquid are not as close together, are arranged in a random way, and can move around each other, while particles in a solid are packed closely together in order and cannot move from place to place.**

43. Give an example from the text that proves different solids melt at different rates or temperatures.

**Answer: If you melt butter in a metal pan, the butter will melt (turn to liquid), but the metal pan will not melt because it requires much more heat.**

44. What must happen to the temperature of a liquid for it to turn into a gas, and what is this process called?

**Answer: The temperature must be increased (heated), and the process is called evaporation.**

45. Why is a change of state (like ice melting) considered a physical change rather than a chemical change?

**Answer: Because it is a reversible change and the matter is not changed chemically into a new product.**

46. What happens to the particles of substances when they go through a chemical reaction?

**Answer: The particles are rearranged to make a completely new product.**

47. Why might a powder, like sugar, be easily confused with a liquid?

**Answer: Because powders can be poured into a container and appear to take its shape, similar to liquids.**

48. Give two examples of chemical reactions mentioned in the text.

**Answer: Digesting food in our bodies, making a baking soda volcano, making plastic from milk, making a bath fizzer, or mixing plaster of Paris with water.**

49. What is the main characteristic of an "irreversible" change?

**Answer: The substances cannot be changed back to their original form because a new product has been made.**

50. Based on the scientific enquiry investigating 'play slime', what questions were scientists trying to answer to classify its state of matter?

**Answer: They asked if it can be compressed, if its shape changes according to the container it is in, and if it always takes up the same amount of space.**